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ENGINES AND ITS APPLICATION TO THE ALLISON V-3420-11 ENGINE

By George F. Kinghorn, Albert H. Schroeder, and William K. Hagginbothom, Jr.

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NATIONAL ADVISORY COMMITTEE FOR AERONAUTICS

# MEMORANDUM REPORT

for the

Army Air Forces, Air Technical Service Command

A METHOD FOR CORRELATING THE COOLING DATA OF LIQUID-COOLED

ENGINES AND ITS APPLICATION TO THE ALLISON V-3420-11 ENGINE

By George F. Kinghorn, Albert H. Schroeder and William K, Hagginbothom, Jr.

#### SUMMARY

A study has been made of the heat-transfer processes in liquid-cooled engines and an equation has been developed that relates the heat rejection to the coolant and the engine operating conditions. Tests of an Allison V=3420-11 engine have been made to check the accuracy of the equation and to establish the cooling characteristics of the engine. By determining the few constants of the equation, the heat rejection to the coolant may be predicted with good accuracy for any particular engine operating condition. The tests showed that the rate of heat dissipation to the coolant was only slightly affected by either the rate of coolant flow or the relative proportions of ethylene glycol and water composing the coolant mixture.

#### INTRODUCTION

An analysis has been made of the heat-transfer processes in liquid-cooled engines to determine the effects of the various engine and cooling parameters upon the heat rejection to the coolant. This analysis parallels the analysis of heat-transfer processes in air-cooled engines presented in reference 1.

In the analysis of reference 1, equations were developed that relate the cylinder temperatures and the engine operating conditions. These equations have proved very useful in providing a means of completely determining the cooling characteristics of air-cooled engines with a

minimum of testing. In the present report somewhat similar equations are developed that give the heat rejection of liquid-cooled engines as a function of the engine operating conditions.

Tests of an Allison V-3420-11 24-cylinder, liquid-cooled engine installed in an XB-39 nacelle were made to check the analysis and to determine the heat-rejection characteristics of this engine. The tests were made over a wide range of engine operating conditions. The coolants used were ethylene glycol, water, and two mixtures of ethylene glycol and water.

# SYMBOLS

- A, A', B, a, b, d, f, m, n constants
- cp specific heat of fluid at constant pressure,
  Btu per pound per OF
- opa specific heat of air at constant pressure, Btu per pound per of
- specific heat of coolant at constant pressure,
  Btu per pound per OF
- specific heat of combustion gases at constant pressure, Btu per pound per OF
- g acceleration due to gravity, feet per second per second
- H rate of heat transfer, Btu per second
- H<sub>c</sub> rate of heat transfer from cylinder walls to coolant, Btu per second
- Hg rate of heat transfer from combustion gases to cylinder walls, Btu per second
- h surface heat-transfer coefficient, Btu per second per square foot per OF
- J mechanical equivalent of heat, foot-pounds per Btu
- k thermal conductivity of fluid, Btu per second per square foot per OF through 1 foot

k <sub>o</sub>	thermal conductivity of coolant, Btu per second per square foot per of through 1 foot
k <sub>g</sub>	thermal conductivity of combustion gases, Btu pe second per square foot per OF through 1 foot
Z	linear dimension of fluid passageway, feet
ន	surface area in contact with fluid, square feet
t <sub>e</sub>	average coolant temperature through engine, OF
tcarb	carburetor-air temperature, <sup>O</sup> F
t <sub>f</sub>	average temperature of fluid, OF
<sup>t</sup> g	effective gas temperature, OF
<sup>t</sup> g <sub>o</sub>	effective gas temperature for 0° F intake-air temperature, °F
t <sub>w</sub>	average cylinder-wall temperature, OF
t <sub>w</sub> '	temperature of cylinder wall measured with embedded thermocouples at locations shown in figure 3, of
V	average velocity of fluid, feet per second
$v_t$	impeller tip speed, feet per second
Wo	coolant flow rate, pounds per second
We	engine-air flow rate, pounds per hour
∆ե <sub>Ծ</sub>	blower temperature rise, OF
ΔT <sub>C</sub>	coolant temperature rise through engine, OF
μ	absolute viscosity of fluid, slugs per second per foot
μο	absolute viscosity of coolant, slugs per second per foot
$\mu_{\mathbf{g}}$	absolute viscosity of combustion gases, slugs per second per foot
^	density of fluid along par subjected

N engine speed, rpm

pm manifold pressure, inches of mercury absolute

$$z = \frac{1}{A \frac{k_c}{\mu_c} \left(\frac{c_{p_c} \mu_c g}{k_c}\right)^{0.14} w_c^d}$$

F correction factor applied to obtain Z (fig. 9(b))

K<sub>r</sub> correction factor for fuel-air ratio (fig. 13)

 $K_N$  correction factor for engine speed (fig. 13)

### ANALYSIS

An understanding of the factors determining the amount of heat rejected to the coolant in a liquid-cooled engine can be obtained from a study of the processes by which heat is transferred from the combustion gases to the cylinder walls and from the cylinder walls to the coolant. It has been shown that nearly all the heat transferred from the combustion gases to the cylinder walls is transferred by forced convection. Heat may be transferred from the cylinder walls to the coolant either by forced convection or, if the temperature of the coolant is sufficiently high, by a combination of forced convection and boiling.

Tests have shown that, in general, moderate boiling or vaporization of the coolant in a liquid-cooled engine has little effect upon the over-all rate of heat transfer. Results of tests of an Allison V-1710-81 engine at the NACA Aircraft Engine Research Laboratory, Cleveland, Chio, indicated that reducing the coolant pressure from 30 to 15 pounds per square inch absolute increased the heat transfer not more than about 3 percent, even though in some cases violent boiling occurred. During the present investigation of the Allison V-3420-11 engine, preliminary tests showed that varying the coolant temperature as much as 80° F resulted in a variation in heat transfer approximately equal to that which would be expected for a forced-convection process. Some evidence indicates that with

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very violent boiling, particularly at low coolant flows, the heat transfer is affected to a fairly large degree. For normal engine operation, however, the effect of moderate boiling may be neglected and in the present report the heat to the coolant will be considered to be transferred entirely by forced convection.

Dimensional analysis has shown and experiment has verified that for heat transfer by forced convection the Nusselt number  $\frac{ht}{k}$  is a function of the Reynolds number  $\frac{\rho vt}{\mu}$  and the Prandtl number  $\frac{c_p \mu g}{k}$ . Test data have indicated that these functions are simple exponential functions for either laminar flow or fully developed turbulent flow; that is,

$$\frac{hl}{k} \propto \left(\frac{pvl}{\mu}\right)^m \left(\frac{c_p \mu g}{k}\right)^n \tag{1}$$

where m and n are constant over fairly wide ranges of Reynolds and Prandtl numbers, except in the transition region between laminar and turbulent flow. The rate of heat transfer is given by the relationship

$$H = hS(t_f - t_w)$$
 (2)

where S is the surface area over which the fluid flows and  $t_f$  and  $t_w$  are the average temperatures of the fluid and wall, respectively. Equations (1) and (2) may be combined to give

$$H \propto S_{\overline{l}}^{\underline{k}} \left(\frac{\rho V l}{\mu}\right)^{\underline{m}} \left(\frac{c_p \mu g}{\underline{k}}\right)^{\underline{n}} \left(t_f - t_{\underline{w}}\right) \tag{3}$$

For the heat-transfer process from the combustion gases to the cylinder walls,  $\iota$  in equation (3) is some representative internal dimension of the cylinder. Since  $\iota$  and S are constant for a particular engine and since the engine-air flow  $\mathbf{W}_{\mathbf{e}}$  is proportional to  $\rho\mathbf{V}$ , the heat transferred from the combustion gases is

$$H_{g} \propto W_{e}^{a} \frac{k_{g}}{\mu_{g}^{a}} \left(\frac{c_{p_{g}}\mu_{g}e}{k_{g}}\right)^{b} \quad (t_{g} - t_{w}) \tag{4}$$

where  $t_g$  is the effective temperature of the gases within the cylinder over the entire cycle and the values of  $k_g$ ,  $\mu_g$ , and  $c_{p_g}$  are also effective values over the entire cycle. It is indicated in reference 1 that the value of the term  $\frac{k_g}{\mu_g} \left( \frac{o_{p_g} \mu_g g}{k_g} \right)^b$  does not vary

appreciably with changes in engine operating conditions. More recent data on the effect of temperature and fuel-air ratio upon the physical properties of mixtures of fuel and air after compustion (references 2)

and 3) indicate that 
$$\frac{k_g}{\mu_g^2} \left(\frac{c_{p_g}\mu_g g}{k_g}\right)^b$$
 may vary appreciably

with changes in engine fuel-air ratio. Since  $t_g$  is also a function of fuel-air ratio, however, the variations  $k_{\perp}$  /on  $u_{\perp}g^{(b)}$ 

of 
$$\frac{k_g}{\mu_g} \left( \frac{o_{D_g} \mu_g g}{k_g} \right)^b$$
 may, to a first approximation, be

included in the effective gas temperature. Equation (4) may therefore be written

$$H_g \propto W_e^a (t_g - t_w)$$
 (5)

where tg may be defined as the temperature that most nearly satisfies equation (5) and is a function of only fuel-air ratio, intake-air temperature, spark timing, and exhaust back pressure. A large number of tests have shown that equation (5) is very accurate for air-cooled engines; this equation may be expected to be equally accurate for liquid-cooled engines.

The rate of heat transfer from the cylinder walls to the coolant is, by a similar analysis,

$$H_{c} \propto W_{c}^{d} \frac{k_{0}}{\mu_{c}^{d}} \left(\frac{c_{p_{c}}\mu_{c}g}{k_{c}}\right)^{f} \left(t_{w} - t_{c}\right)$$

or

$$H_{c} = AW_{c}^{d} \frac{k_{c}}{\mu_{c}^{d}} \left(\frac{c_{p_{c}}\mu_{c}g}{k_{c}}\right)^{f} (t_{w} - t_{c})$$
 (6)

The value of the term  $\frac{k_c}{\mu_c d} \left(\frac{c_{p_c} \mu_c g}{k_c}\right)^r$  in this case depends

upon the proportions of ethylene glycol and water used as the coolant and upon the coolant temperature.

Because of the heat generated by friction between the piston rings and the cylinder barrels, the cooling of the pistons and barrels by the oil, and the cooling of the exposed surfaces of the cylinder block, the heat transferred from the combustion gases to the cylinder walls H<sub>c</sub> is not equal to the heat transferred to the coolant H<sub>c</sub>. It is assumed, however, that

$$H_{c} \propto H_{g}$$

Therefore, from equation (5),

$$H_{c} = BW_{e}^{a} \left(t_{g} - t_{w}\right) \tag{7}$$

or

$$t_w = t_g - \frac{H_c}{BW_c}$$

Substituting  $t_g - \frac{H_o}{BW_e}$  for  $t_w$  in equation (6) yields

$$H_{c} = AW_{c}^{d} \frac{k_{c}}{\mu_{c}^{d}} \left(\frac{c_{c}c^{\mu_{c}g}}{k_{c}}\right)^{f} \left(t_{g} - t_{c} - \frac{H_{c}}{BW_{e}^{g}}\right)$$

and

$$\frac{t_g - t_o}{H_c} = \frac{1}{BW_e^a} + \frac{1}{A \frac{k_o}{\mu_c^d} \left(\frac{c_{p_c} \mu_c g}{k_c}\right)^T W_c^d}$$
(8)

Equation (8) is used in the present report to correlate the data obtained on the V-3420-11 engine.

For a single coolant mixture and coolant temperature, the physical properties of the coolant are constant and equation (8) may be rewritten as

$$\frac{\mathbf{t_g - t_c}}{\mathbf{H_c}} = \frac{1}{\mathbf{BW_e}^a} + \frac{1}{\mathbf{A'W_c}^d} \tag{9}$$

#### ENGINE AND INSTRUMENTATION

The Allison V-3420-11 engine tested is a 24-cylinder, double-vee, liquid-cooled engine, having a normal-power rating of 2100 horsepower at 2600 rpm and a military-power rating of 2600 horsepower at 3000 rpm. The engine has a compression ratio of 6.65:1, a propeller-gear ratio of 3.13:1, and a blower-gear ratio of 6.9:1. The impeller diameter is 10 inches. The engine is equipped with a Bendix-Stromberg PR58B3 carburetor. A view of the engine nacelle as set up for the tests is shown in figure 1.

The coolant system for this engine installation had two radiators, one for each half of the engine. The coolant flow from each half of the engine was measured by an annular-orifice flowmeter. A close agreement with these flow measurements was obtained from pitot-static

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flowmeters that measured the flow into the individual cylinder blocks.

The coolant temperature rise was measured by three thermocouples across each half of the engine. The hot and cold junctions of these thermocouples were located in the coolant piping near the engine outlet and inlet, respectively. A hand-balanced potentiometer was used to indicate the coolant temperature difference existing between the hot and cold junctions. In general, the coolant temperature rise through the engine is difficult to measure accurately. Some of the difficulty is due to the fact that the over-all temperature difference is small throughout the engine operating range. Limitations in space available for the installation of thermocouples present an additional practical problem. accuracy of the method used in these tests is estimated to be the percent. The temperature of the coolent entering the engine was measured by a resistance thermometer in conjunction with a special microammeter.

A sketch of part of the coolant system, which shows the points of flow and temperature measurement for the left half of the engine, is given as figure 2. The instrumentation for the coolant system on the right half of the engine was similar to that shown for the left half.

Cylinder temperatures were measured by embedded thermocouples and spark-plug-gasket thermocouples. Embedded thermocouples were located between the intake valves, between the exhaust valves, and in the exhaust spark-plug bosses (as shown in fig. 3) of cylinders 1, 2, 3, 4, 5, and 6 of the left bank and 1, 3, and 6 of each of the other three banks. Cylinder-barrel temperatures were not measured because of the difficulty of installing thermocouples. The carburetor-air temperature was measured by two thermocouples soldered to the carburetor screen.

The engine-air flow was measured by a calibrated venturi (fig. 1). Fuel flow was measured by rotameters and a weigh tank. Standard aircraft instruments were used to measure manifold pressure and engine speed. A special mercury U-tube manometer was also used to measure manifold pressure for some of the tests.

# METHODS AND TESTS

Tests were made with the following four coclant mixtures: (a) 100 percent ethylene glycol (AN-E-2), (b) 80 percent by volume ethylene glycol (AN-E-2) and 20 percent water, (c) 30 percent by volume ethylene glycol (AN-E-2) and 70 percent water, and (d) water.

In order to reduce coolant boiling during the tests with the 30-70 mixture and with water, the coolant system was pressurized by applying compressed air to the expansion tank. A sight glass was installed to indicate the coolant level. No appreciable increase in the coolant level was observed during any of the tests - an indication that large vapor pockets did not form.

The properties of the coolants were obtained from reference 4. Curves showing these properties for mixtures of pure ethylene glycol and water have been plotted from the data of reference 4 and are presented in figure 4. Ethylene glycol (AN-E-2) was considered to be 97 percent by volume pure ethylene glycol and 3 percent water, with the effect of the inhibitor neglected.

The heat rejection to the coolant was determined by use of the following equation:

$$H_{c} = W_{c}c_{p_{c}} \Delta T_{c}$$
 (10)

where  $\Delta T_{c}$  is the temperature rise of the coolant measured across the engine.

The constant f in equation (6) was assumed to equal 0.4 as found by Sherwood and Petrie (reference 5). The constants A and d were determined from a plot of

$$\frac{F_{c}}{k_{c}\left(\frac{c_{p_{c}}\mu_{c}g}{k_{c}}\right)^{0.4}(t_{w}-t_{c})}$$

against  $W_{\text{c}}/\mu_{\text{c}}$  on logarithmic paper. The term  $t_{\text{w}}$  -  $t_{\text{c}}$  was found from the equation

$$t_w - t_c = 0.64(t_w' - t_c)$$
 (11)

where tw' is the average temperature measured by the thermocouples embedded in the cylinder block between the intake valves, between the exhaust valves, and in the exhaust spark-plug bosses. The factor 0.64 was obtained from unpublished data from tests at the Cleveland Laboratory of an Allison V-1710 engine in which temperatures were measured at various other points on the cylinder in addition to those between the valves and in the exhaust spark-plug boss. It was found that equation (11) holds closely for all operating conditions. Inasmuch as the cylinders of the V-3420-11 and V-1710 engines are nearly identical, it may be expected that this relationship is also valid for the V-3420-11 engine.

No tests were made to determine the effects of spark timing, exhaust back pressure, or intake-air temperature upon tg. There is no provision on the V-3420-11 engine for varying the spark timing and all the tests were made with normal spark timing. The exhaust back pressure was approximately 30 inches of mercury absolute throughout the tests. The results are not applicable, therefore, at high altitude except for a turbosupercharger installation with the engine operating at high powers.

It has been assumed that, as was found for the cylinder head of an air-cooled engine (reference 6), tg increases approximately 0.8° per degree rise in intake-air temperature; that is,

$$t_g = t_{g_o} + 0.8(t_{carb} + \Delta t_b)$$

where  $t_{carb}$  is the carburetor-air temperature and  $\Delta t_b$  is the blower temperature rise. The blower rise was calculated from the following equation:

$$\Delta t_b = \frac{v_t^2}{c_{p_a} Jg}$$

where  $c_{pa}$  is the specific heat of air at constant pressure and  $V_t$  is the impeller tip speed. For the V-3420-11 engine, this equation may be written

*:* \_

$$\Delta t_{\rm b} = 0.0000151N^2$$

where N is engine speed in revolutions per minute.

In order to determine the actual value of  $t_{S_0}$  at a fuel-air ratio of 0.08, tests were made at constant engine operating conditions and varying coolant temperatures. Since, with constant  $w_{\rm e}$ ,

$$\propto t_{g_0} + 0.8(t_{carb} + \Delta t_b) - t_w$$

 $t_{g_0}$  was found by plotting  $t_w - 0.8(t_{carb} + \Delta t_b)$  against  $H_c$  and extrapolating the resulting curves to  $H_c = 0$  for which

$$t_{g_0} = t_w - 0.8(t_{carb} + \Delta t_b)$$

The values of  $t_w$  used were obtained from the equation

$$t_{w} = \frac{H_{c}}{A \frac{k_{c}}{\mu_{c}} \left(\frac{c_{p_{c}} \mu_{c} g}{k_{c}}\right)^{0.14} W_{c}^{d}} + t_{c}$$

The coolant mixture used for these tests was 100 percent ethylene glycol (AN+E-2).

Tests were made at various fuel-air ratios with engine speed and engine-air flow held constant to determine the value of  $t_{g_0}$  at fuel-air ratios other than 0.08. With  $t_c$ ,  $w_c$ , and  $w_e$  held constant, the value of  $t_{g_0}$  could be found for the different fuel-air ratios from the

equation

$$\frac{\mathbf{t_g} - \mathbf{t_o}}{\mathbf{H_c}} = \mathbf{Constant}$$

The value of the constant was determined from the measured values of  $t_c$  and  $H_c$  at a fuel-air ratio of 0.08 and the value of  $t_{g_0} + 0.8(t_{carb} + \Delta t_b)$  previously determined for a fuel-air ratio of 0.08.

Tests were made at various engine speeds, engine powers, and fuel-air ratios with each of the four coolants. The value of

$$\frac{\mathbf{t_g - t_c}}{\mathbf{H_c}} - \frac{1}{\mathbf{A} \cdot \frac{\mathbf{k_c}}{\mu_c^d} \left(\frac{\mathbf{c_{p_c} \mu_c g}}{\mathbf{k_c}}\right)^f \mathbf{w_c}^d}$$

was determined for each test and plotted against  $\mathbf{W}_{\mathbf{e}}$  on logarithmic coordinates.

# RESULTS AND DISCUSSION

$$\frac{\mathbf{H_c}}{\mathbf{k_c} \left(\frac{\mathbf{c_{p_c}} \mu_o \mathbf{g}}{\mathbf{k_c}}\right)^{0.14} (\mathbf{t_w} - \mathbf{t_c})}$$

against  $W_0/\mu_C$  from which the values of d and A of equations (6) and (8) were determined is shown in figure 5. It may be seen that the slope d of the curve of figure 5 is not constant but decreases with increasing  $W_0/\mu_C$ , as is usually observed for the transition region between laminar and fully developed turbulent flow.

Separate values of d and A were selected from figure 5 for each of the coolants tested. These values are as follows:

Coolant mixt (percent by vo			
Ethylene glycol (AN-E-2)	Water	đ	A
100	0	0.34	309
80	20	.28	728
30	70	.17	3,750
0	100	.095	11,600

The value of t<sub>80</sub> at a fuel-air ratio of 0.08 was determined from the plot shown in figure 6. On an average the data indicate that t<sub>80</sub> for the entire cylinder is approximately 700° F. Because of the large extrapolation necessary in figure 6, values of 600° F and 800° F for t<sub>80</sub> at a fuel-air ratio of 0.08 were also used in calculating the test data. Closer correlation between the heat rejection to the coolant and the engine operating conditions was obtained by using 700° F than by using either 600° F or 800° F. The data obtained by using 600° F and 800° F are not given in the present report.

A value of approximately 900° F for tgo for the entire cylinder of an air-cooled engine was calculated from data given in reference 1 for a Pratt & Whitney R-1340-H cylinder and in reference 6 for a Wright R-1820-G cylinder. The reason for the lower value of for the V-3420-11 cylinder is  $t_{g_o}$ 700° F obtained for not entirely understood but this lower value may in part be due to better scavenging of the V-3420-11 cylinder, which has two intake and exhaust valves with a comparatively large valve overlap of 65°. Differences in cylinder construction, compression ratio, or spark timing may also have contributed to the differences in tgo between these engines.

The effect of fuel-air ratio upon  $t_{g_0}$  is shown in figure 7. Figure  $\delta$  shows

$$\frac{\mathbf{t_g - t_o}}{\mathbf{H_o}} - \frac{1}{\mathbf{A} \frac{\mathbf{k_o}}{\mu_o d} \left(\frac{\mathbf{o_{p_c} \mu_c g}}{\mathbf{k_c}}\right)^{0.4} \mathbf{w_c}^d}$$

plotted against  $W_e$ . Some of the scatter in figure 8 can be attributed to the limited accuracy of the method used to measure the coolant temperature rise.

The heat-transfer equations obtained from equation (8) and figures  $\delta$  and 5 are as follows:

Coolant		
(percent by Ethylene glycol (AN-E-2)	Water	Heat-transfer equation
100	0 .	$\frac{\frac{t_g - t_o}{H_c}}{309 \frac{k_c}{\mu_o^{0.34}} \left(\frac{c_{p_o} \mu_c g}{k_o}\right)^{0.44} \pi_o^{0.34}} = 135\pi_e^{-0.52}$
80	20	$\frac{\frac{t_g - t_o}{H_o}}{728 \frac{k_c}{\mu_o^{0.28}} \left(\frac{c_{p_o} \mu_o g}{k_c}\right)^{0.14} w_o^{0.28}} = 135 m_e^{-0.52}$
30	70	$\frac{t_{g}-t_{o}}{H_{c}} = \frac{1}{3.750 \cdot \frac{k_{o}}{\mu_{o}^{0.17}} \left(\frac{c_{p_{o}}\mu_{o}g}{k_{o}}\right)^{0.17}} = 135W_{e}^{-0.52}$
	100	$\frac{\frac{t_g - t_o}{H_o}}{11.600 \frac{k_o}{\mu_o^{0.095}} \left(\frac{c_{p_o} \mu_o g}{k_o}\right)^{0.14} \pi_o^{0.095}} = 135 \pi_o^{-0.52}$

The values of the second term of these equations

$$\frac{1}{A \frac{k_c}{\mu_c^d} \left(\frac{c_{p_c} \mu_c g}{k_c}\right)^{0.4} W_c^d}$$

for various coolant mixtures, coolant temperatures, and coolant flow rates are presented in figure 9. For convenience in the use of these curves,

$$\frac{1}{A \frac{k_c}{\mu_c^d} \left(\frac{c_{p_c} \mu_c g}{k_c}\right)^{0.4} \psi_c^d}$$

is denoted by Z in figure 9.

The small effect of changes in coolant flow rate and coolant properties upon the engine heat rejection may be seen from the foregoing heat-transfer equations. The value of the term

$$\frac{1}{309 \frac{k_{c}}{\mu_{c}^{0.34}} \left(\frac{c_{p_{c}}\mu_{c}g}{k_{c}}\right)^{0.4}} W_{c}^{0.34}$$

for 100 percent ethylene glycol (AN-E-2) at normal power is approximately 0.20, whereas  $\frac{t_g-t_c}{H_c}$  is approxi-

mately 1.15. An increase in coolant flow W<sub>c</sub> of 50 percent results in an increase in heat transfer of only 2.5 percent if other conditions remain constant. A change in coolant to a mixture of 30 percent ethylene glycol (AN-E-2) and 70 percent water results in a value

of--

$$\frac{1}{3750 \frac{k_0}{\mu_0^{0.17}} \left(\frac{c_{p_c}\mu_0g}{k_c}\right)^{0.17}} W_0^{0.17}$$

of approximately 0.145, which would cause an increase in heat transfer of about 5 percent.

The variation in average cylinder-wall temperature  $t_{\rm w}$  with changes in coolant flow rate or coolant properties may be found from the resulting change in  $H_{\rm c}$  and from the equation

$$H_c = BW_e^a(t_g - t_w)$$

If  $W_e$  is constant,

$$\mathbf{H_c} \propto \mathbf{t_g} - \mathbf{t_w}$$

Because of the small effect upon the heat rejection of variations in

$$\frac{1}{\mathbb{A} \frac{k_{0}}{\mu_{c}^{d}} \left(\frac{c_{p_{c}}\mu_{c}g}{k_{c}}\right)^{0.14} \mathbf{W}_{c}^{d}}$$

with changes in coolant flow rate  $w_c$  and average engine coolant temperature  $t_c$ , the test data for individual coolant mixtures may be plotted as shown in figure 10.

By plotting  $\frac{H_C}{t_c}$  against  $W_e$ , curves are obtained from which the heat rejection  $H_C$  may be determined with only a small sacrifice in accuracy much more easily than from figure S.

It was assumed in the analysis that the heat generated by friction between the piston rings and the cylinder barrels has little effect upon the heat rejection to the coolant. Tests were made with constant engine-air consumption at various engine speeds to determine the error involved in the use of this assumption. The data indicated that

$$\frac{\mathbf{t_g - t_c}}{\mathbf{H_c}} - \frac{1}{\mathbf{A} \frac{\mathbf{k_c}}{\mu_c^d} \left(\frac{\mathbf{c_{p_c} \mu_c g}}{\mathbf{k_c}}\right)^{0.14} \mathbf{W_c}^d}$$

decreases slightly with increasing engine speed, but an accurate evaluation of this effect was not possible primarily because of the limited accuracy of the method used to measure the coolant temperature rise.

In order to relate the cooling characteristics of the engine to variables measured by the usual engine instruments, calibration curves of  $W_{c}$  and  $W_{e}$  are presented in figures 11 and 12, respectively. Figure 11 shows that the proportions of water and ethylene glycol used for the coolant have little effect upon the rate of coolant flow. If cavitation occurs at the coolant pump inlet, however, the flow may be considerably less than that shown in figure 11. In preparing figure 12, the engine-air flow  $W_{e}$  was assumed to be a function of only the engine speed, manifold pressure, exhaust back pressure, and the sum of the absolute carburetor-air temperature and the blower temperature rise  $t_{carb} + 460 + \Delta t_{b}.$  Data from the Allison Division of General Motors Corp. indicate that the engine-air flow varies inversely as  $\sqrt{t_{carb} + 460} + \Delta t_{b}.$  Curves of

$$\frac{w_{e}}{\sqrt{t_{carb} + 460 + \Delta t_{b}}}$$
 against engine speed are plotted in

figure 12 for various manifold pressures. Throughout the tests, the maximum difference between the measured engine-air flow and the corresponding values given by figure 12 was less than 4 percent. No data were available concerning the effect of exhaust back pressure upon engine-air flow.

The variation of brake horsepower with engine operating conditions may be determined from the following empirical relation obtained from the Allison Division:

where

pm manifold pressure, inches of mercury absolute

t<sub>carb</sub> carburetor-air temperature, <sup>o</sup>F

 $K_N$  correction factor for engine speed (fig. 13)

Kg correction factor for fuel-air ratio (fig. 13)

The data obtained during the tests are presented in table I.

# APPLICATION

Through the use of the curves presented in the present report, the heat rejection to the coolant for the Allison V-3420-11 engine may be determined for any particular engine operating condition. The following example, based on engine operation at military power (2600 bhp at 3000 rpm and a manifold pressure of 44.5 in. of mercury absolute), illustrates the procedure: Typical operating conditions assumed are

For engine operation at 3000 rpm and 44.5 inches of mercury absolute, figure 12 indicates that

$$\frac{W_e}{\sqrt{t_{carb} + 460 + \Delta t_b}} = 697$$

The blower temperature rise is
$$\Delta t_b = \frac{v_t^2}{c_{p_a} J_g}$$
= 0.0000151 $n^2$ 
= 0.0000151(3000)<sup>2</sup>
= 135.9° F

Solving for the engine-air flow yields

$$W_{e} = 697\sqrt{t_{carb} + 460 + \Delta t_{b}}$$
$$= 697\sqrt{80 + 460 + 136}$$

= 18,100 pounds per hour

For an engine-air flow of 18,100 pounds per hour (fig. 8),

$$\frac{t_{g} - t_{c}}{H_{c}} - \frac{1}{A \frac{k_{c}}{\mu_{c}^{d}} \left(\frac{c_{p_{c}} \mu_{c} g}{k_{c}}\right)^{0.14}} = 0.820$$

In order to determine the value of

$$\frac{1}{A \frac{k_c}{\mu_c d} \left(\frac{c_{p_c} \mu_c g}{k_c}\right)^{0.4} w_c^d}$$

it is necessary to know the coolant flow rate  $W_{\rm C}$  and the average engine coolant temperature  $t_{\rm C}$ . From figure 11,  $W_{\rm C}=78$  pounds per second at 3000 rpm. It was found during the tests that the average engine coolant temperature was approximately 5° F lower than the coolant temperature out of the engine over a wide range of operating conditions; therefore, let  $t_{\rm C}=245^{\circ}$  F. Then, from figure 9(a), for a coolant mixture of 70 percent by volume ethylene glycol (AN-E-2), a coolant flow rate of 78 pounds per second, and an average engine coolant temperature of 250° F, the term

$$\frac{1}{A \frac{k_c}{\mu_c^d} \left(\frac{c_{p_c}\mu_c g}{k_c}\right)^{0.14} w_c^d}$$

denoted by  $Z_{(t_c=250)}$  is equal to 0.160. In order to correct this term to the desired value of  $t_c$ ,  $245^{\circ}$  F, for the same  $W_c$  and coolant mixture, the correction factor F in figure 9(b) is found to be 0.993. Therefore,

$$Z = FZ(t_c=250)$$
  
= 0.993 × 0.160  
= 0.159

Then

$$\frac{t_g - t_o}{H_o} - 0.159 = 0.820$$

or

$$H_{c} = \frac{t_{R} - t_{c}}{0.979}$$

For a fuel-air ratio of 0.095,  $t_{g_0} = 649^{\circ}$  F from figure 7.

Then

$$t_g = t_{g_0} + 0.8(t_{carb} + \Delta t_b)$$

$$= 649 + 0.8(80 + 136)$$

$$= 822^{\circ} F$$

Since  $t_c = 245^\circ F$ ,

$$H_{c} = \frac{t_{g} - t_{c}}{0.979}$$
$$= \frac{822 - 245}{0.979}$$

= 589 Btu per second

The Allison Division guarantees that the heat rejection to the coolant at military power shall not exceed 608 Btu per second, which is approximately 3 percent above the heat rejection calculated in the preceding example.

# CONCLUSIONS

As a result of an analysis made of the heat-transfer processes in liquid-cooled engines, an equation has been developed that relates the heat rejection to the coolant and the engine operating conditions. Tests of an Allison

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V-3420-11 engine over a wide range of operating conditions and for several coolant mixtures showed that:

- 1. By determining the constants of the equation, the heat rejection to the coolant may be predicted with good accuracy for any particular operating condition.
- 2. The rate of coolant flow had only a slight effect upon the rate of heat dissipation to the coolant; also, the effect of the relative proportions of ethylene glycol and water composing the coolant mixture upon the heat-dissipation rate was small.
- 3. Changes in engine friction with engine speed had a small effect upon the heat rejection to the coolant; an accurate evaluation of this effect was not made.
- h. The effective gas temperature for an entire cylinder of the V-3 $\frac{1}{2}$ O-11 engine was approximately 700° F for a fuel-air ratio of 0.08 and an intake-air temperature of 0° F.

Langley Memorial Aeronautical Laboratory
National Advisory Committee for Aeronautics
Langley Field, Va.

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[Engine serial no. h2-271081; compression ratio, 6.65 to 1; spark timing, intake 28° B.T.C., exhaust 34° B.T.C.; carburstor, Bendix-Stromberg PR58B3; fuel, AN-F-28; oil, AN-VV-0-446, grade 1120]

			Carburetor- air tem- perature (°F)				low air				Average		Cylinder-wall temperature (°F)			
Test	Manifold pressure (in. Hg abs.)	Engine speed (rpm)		Barometric pressure (in. Hg abs.)	Fuel flow (lb/hr)	Engine- air flow (1b/hr)		system	tem tempera- ture into engine	tock   Dt =hel	engine	Engine coolant flow (lb/sec) Left Right Total	Embedded thermo- souple	Spark-plug- gasket thermo- couple	Engine heat rejection (Btu/sec)	Oil tempera- ture into engine (OF)
								III. Eago,	(-1)	half half Av.		half half Total	Av. Max.	Av. Max.	<u> </u>	
	<u></u>		<del></del>	,		C	olant	mixture,	80 percent	by volume ethyle	ne glycol		<del></del>	· 	····	
127456789011274567990122745678901274567990127456799012	00000000000000000000000000000000000000	1900 1900 1900 1900 1900 1900 1900 1900	996777888888888888888888889998804510965557000344667991999555577	30.1642233455645323345560005555600055556000555600055560005556000555600055560005556000555600055560005556000555600055560005556000555600055560005000055600055600055600055600055600055600055600055600055600055600005560005560005560005560005560005560005560005560005560005560005560000500005000050000500005000050000050000	3594176222LBB04555827743274545454545500872588574665138888566651388885666513888856665138888566651388888	7050 7100 7100 7100 7100 71100 71100 71100 71100 16100 1620 16700	.06736 .067328 .067328 .07997 .07997 .07997 .0801 .0801 .0807 .0801 .0807 .0806 .0807 .0806 .0807 .0808 .080		207 206 204 212 201 206 206 206 205 205 207	10.3   9.9   10.1   10.5   10.3   10.4   10.0   10.6   10.5   10.5   10.4   10.5   10.2   11.2   10.7   10.3   11.6   11.0   11.3   10.5   10.4   11.1   11.5   11.7   11.6   10.6   11.1   11.1   11.5   11.6   11.6   11.6   11.6   11.6   11.6   11.6   11.7   11.0   11.7   11.6   11.6   11.6   11.7   11.0   11.7   11.8   11.9   11.7   11.9   11.0   11.5   11.9   11.0   11.5   11.9   11.0   11.5   11.9   11.0   11.5   11.9   11.0   11.5   11.0   10.5   10.6   11.9   11.5   11.6   11.1   11.5   11.7   11.6   11.6   11.6   11.7   11.8   11.7   11.6   11.6   11.6   11.7   11.8   11.7   11.8   11.8   11.7   11.9   11.1   11.5   11.1   11.5   11.5	212 229 229 229 220 220 220 220 210 200 210 209 210 209 210 209 210 209 210 211 212 229 210 211 212 229 210 211 212 229 210 211 212 229 210 211 212 229 210 211 212 229 210 211 211 212 229 210 210 211 212 229 210 210 211 212 229 210 210 210 210 210 210 210 210 210 210	21. 4 6.6 6.6 6.6 6.6 6.6 6.6 6.6 6.6 6.6	159 159 159 159 159 159 159 159 159 159	272 297 281 3030 273 3000 273 3000 273 3000 273 3000 273 3000 273 3000 273 297 281 2868 264 2885 264 2885 265 2881 266 218 22776 266 2281 266 2281 267 2681 268 2681 2681 2681 2681 2681 2681 2681 2681	5576638998322754817029644822193186155783102252345592235265 557788893223555223559600000000000000000000000000000000000	166777766155555555564884422277355577718888880159778241128808115651166555555555555644811662277355577188888811558241128808116651166511665116651166511665116651

NATIONAL ADVISORY COMMITTEE FOR AERONAUTICS

TABLE I	RESULTS	OF 1	TESTS (	ON	ALLISON	V-3420-11	ENGINE -	Continued

							D. al	Garlant		coolant Engine coolan			Average	Engine coolant					er-wall rature F)	Engine	011
Test	Manifold pressure (in. Hg abs.)	Engine speed (rpm)	Carburetor- air tem- perature (°F)	Barometric pressure (in. Hg abs.)	Fuel flow (lb/hr)	Engine- air flow (1b/hr)	Fuel- air ratio	aystem	Coolant tempera- ture into engine (OF)	temp	eratur (°F) Right	re rise coolant tempera- ture		()	flow lb/sec Right	)	th	edded ermo- uple	Spark-plu gasket thermo- couple	Engine heat rejection (Btu/sec)	011 tempera- ture into engine (OF)
										half	half			half	half	ا	٨٧.	Max.	Av. Max.		
5545567890012345666890	24.1.100.1.2.00.1.0.2.3.3.3.3.2.1.2.1.2.0.0.0.0.0.0.0.0.0.0.0.0.0.0.0	1500 1710 2590 2490 2250 2150 1105 1300 1005 1200 2200 2200 2200 2200 2200	76 77190 8887000 88888888888888888888888888888	30.27 30.26 30.15 30.19 30.09 30.06 30.07 30.06 30.07 30.06 30.04 30.04	295 403 13001 845 7477 1226 910 4810 1804 5684 5684 5684 5684 5684 5684	\$250 5500 15280 11190 10100 9310 6680 18100 2860 1100 2710 8710 8710 8700 8700 8700 8700	0.0694 .0733 .0983 .0972 .0872 .0876 .0787 .0712 .0867 .0900 .0851 .0907 .0675 .0671 .0676		205 203 205 205 205 205 205 205 205 205 205 205	11.9 11.6 10.9 11.0 11.3 11.8 12.1 10.6 7.7 9.1 10.3 11.0 10.3 11.0 10.3	10.8 10.1 10.5 10.3 11.1 9.3 11.0 9.3 11.1 12.5	11.9 10.5 10.7 10.8 11.4 11.7 9.2 10.8 11.0 11.1 11.0 11.1 11.0 10.2	211 211 209 210 209 210 210 210 211 209 211 209 211 209 209 210 209 209	192.35.66.67.62.77.76.66.66.44.63.77.76.66.66.44.69.66	17.50 20.21 20.21 27.51 22.26 111.60 113.60 113.60 113.60 113.60 27.26 26.89 26.89 26.89	92871846614966666666666666666666666666666666	3355476 3255 3555 3556 3556 3556 3556 3556 355	374 393 439 4431 4520 307 2324 415 415 415 415 417 411	261 288 261 299 282 313 281 523 279 3310 278 330 277 2257 219 241 230 227 267 332 279 336 279 336 279 322 279 222 272 282 283 287	3555 5555 4769 4115 2571 267 4416 4416 4416 4416 4416 4416 4416 44	161 165 165 165 165 165 167 167 160 162 163 163 164 179
						Co		mixture, 1	00 percent	by v	٠	ethyle	ne glycol				L	٠ <u>.                                    </u>	<u> </u>		
71 72 74 76 778 798 80 81 82 84 85 86 87 90 91 92 93 99 90 100 101 102 103	37.00 35.11 32.11 30.11 30.10 30.00 17.00 18.00 17.00 18.00 17.00 18.00 17.00 18.00 17.00 18.00 17.00 18.00 17.00 18.00 17.00 18.00 17.00 18.00 17.00 18.00 17.00 18.00 17.00 18.00 17.00 18.00 17.00 18.00 17.00 18.00 17.00 18.00 17.00 18.00 17.00 18.00 19	2600 2100 21700 2150 2000 2000 2000 2000 2600 1200 2500 1200 2200 1200 2200 1200 2200 1200 200 200	8449142244244400333784477536523309133366	29.75 29.774 29.774 29.774 29.775 29.775 29.886 29.98 29.98 29.98 29.98 29.98 29.98 29.98 29.98 29.98 29.98 29.98 29.98 30.19 30.02 29.886 29.98 30.20 29.886 30.20 29.886 30.20	1258 1058 1058 1059 1059 1059 1059 1059 1059 1059 1059	13110 13510 15430 3780 9280 7970 7970 10010 1650 1640 1650 13430 107940 1120 12210 2310 6260 6270 6260 6270 6260 6270 6260 9440	0919 0707 0707 0707 0690 0708 0692 0809 0809 0809 0809 0716 0716 0716 0716 0716 0716 0716 0716		186 2016 2016 2014 2014 2017 2014 2017 2017 2017 2017 2017 2017 2017 2017	12.1 13.2 12.1 12.3 12.7 12.7 12.7 11.8 13.6 12.7 10.2 11.1 12.0 9.9 9.9 9.8 10.1 10.1 11.1 11.9 10.1 11.9 11.1 11.9 11.9	12.40 111.55 10.68 13.84 10.99 110.99 13.84 19.99 10.90 10.90 10.90 10.90 10.90 10.90 10.90 10.90 10.90 10.90 10.9	11. 8 11. 8 12. 0 12. 0 12. 0 12. 5 12. 5 12. 5 12. 5 12. 5 10. 9 11. 4 12. 5 10. 9 11. 4 12. 5 10. 9 9. 1 11. 6 12. 7 10. 0 10. 0 1	210 2210 2211 2211 2212 2211 222 210 2210 2209 2208 2211 207 208 210 209 212 212 212 212 212 212 212 213 210 209 211 209 209 211 212 212 212 212 213 214 215 216 217 209 209 209 209 209 210 210 210 209 209 209 209 209 209 209 209 209 20	26.4 22.1 19.4 19.3	28.54.94.10.10.77.10.10.77.10.10.77.10.10.77.10.10.77.10.10.77.10.10.77.10.10.77.10.10.77.10.10.77.10.10.10.70.10.10.10.10.10.10.10.10.10.10.10.10.10	61.18.51.03.88.86.51.56.52.1.56.1.17.85.1.17.86.1.17.17.86.1.17.86.1.17.86.1.17.86.1.17.86.1.17.86.1.17.86.1.17.86.1.17.86.1.17.86.1.17.86.1.17.86.1.17.86.1.17.86.1.17.86.1.17.86.1.1	5557479657722 25557479657722 255575127 265575127 2557517 25575127 2557517 25575127 2	463 463 452 388 452 452 453 453 453 302 531 302 531 302 531 302 531 531 531 531 531 531 531 531 531 531	366 366 363 361 367 367 367 367 367 367 367 367	467 467 4681 4495 44168 421 470 470 470 470 470 470 470 470 470 470	166 1654 1765 1645 1659 1659 1758 1664 1659 1664 1659 1659 1659 1659 1659 1659 1659 1659

							Fuel- Coolant   Engine coolant   Engine   Engine						r-wall ature								
Test	Manifold pressure (in. Hg abs.)	Engine speed (rpm)	Carburetor- air tem- perature (OF)	Barometric pressure (in. Hg abs.)	Fuel flow (lb/hr)	Engine- air flow (1b/hr)	Fuel- air ratio	Coolant- system pressure (1b/sq	Coclant tempers- ture into engine			lant e rise	engine coolant tempera-	1	flow (1b/sec)		Embedded thermo-		Spark-plug gasket thermo-	Engine heat rejection (Btu/sec)	Oil tempera- ture into engine (°F)
			, .,					in. gage)	engine (°F)	Left half	Pight half	Av.	(°F)	Left half	Right half	Total	Av. 3	$\rightarrow$	Av. Max.	┤	(°F)
-	Coolant, water													<u> </u>							
105 106 107 108 109 110 111 112 113 114 115 116 117 118 119 120 121	22.1 26.1 28.0 17.0 30.0 34.1 360.0 31.1 360.0 31.1 24.1 24.1	1600 2000 2205 1200 2400 1400 2400 2400 2500 2000 2000 1800 1800 1805 1805	82 861 933 931 96 903 854 87 888 888 94	30.15 30.18 30.18 30.17 30.11 30.18 30.18 30.18 30.18 30.18 30.18	280 475 576 161 123 712 214 1037 1260 692 692 691 432 432 432	3830 6480 79940 12900 93030 10880 11140 8040 8040 8050 8050 8050 8050	0.0751 .07521 .07521 .0854 .0756 .0756 .0756 .0951 .0866 .0866 .0866 .0852	10 5 14 13 	178 176 176 178 177 181 178 192 195 195 187 176 186 186 195 202	8588247129778852887777888528877	77887667777777	88876887778887777	182 180 180 182 181 185 182 198 199 199 191 180 182 199 206	21.3 26.9 30.0 15.8 12.9 33.7 18.4 37.8 27.3 27.3 27.2 24.1 24.1 24.1 24.3	29.8	412.830.055.12.80.07.56.72 65.33.26.66.72	7890760 0 5755 5755 5755 5755 5755 5755 575	55 47 119 122 128	211 230 223 245 229 248 202 216 195 207 201 220 210 262 210 263 236 258 237 259 230 249 220 241 211 233 217 237 225 242 236 248	34518 4518 522716 5203 1720 5493 14368 4266 5555 5555	169 166 172 172 163 165 170 161 161 163 163 161 165 160
	Coolant mixture, 30 percent by volume ethylene glycol (AN-E-2)												L								
122 123 124 126 126 127 126 130 131 132 133 134 135 136 139 141	20.0 20.0 20.0 32.1 34.1 36.0 28.1 21.0 24.2 26.1 30.0 30.1 30.1 30.1 18.0 22.0	11:00 11:00 12:00 26:00 25:00 25:00 18:00 18:00 18:00 21:00 21:00 21:00 21:00 21:00 18:00	92 933 812 807 725 557 22 88 87 725 88 88 88 88 88 88 88 88 88 88 88 88 88	30.08 30.07 30.17 30.18 30.19 30.18 30.19 30.19 30.10 30.11 30.17 30.10 30.11 30.11 30.11 30.11 30.11	213 212 940 940 1618 1280 600 359 360 468 768 765 765 765 745 745	2800 2800 10910 11110 112340 1	.0761 .0757 .0861 .0865 .0968 .0762 .0721 .0726 .0759 .0810 .0810 .0810 .0810	L	177 195 195 195 196 196 196 196 197 212 177 195 213 196 194 195 195	9858 90 99 98 99 88 88 7 6	CO C	962090786411436589	181 199 217 200 199 200 200 200 200 216 182 200 182 217 200 199 199	L	33.6.7.2.9.8.2.3.3.9.5.0.7.5.6.1.5.7.2.2.2.2.3.3.3.0.9.4.1.5.7.1.5.7.2.2.2.2.3.3.3.3.3.2.4.1.5.7.2.2.2.3.3.3.3.3.3.3.3.3.3.3.3.3.3.3.3	98768167233565719563 334937766662346341904 44566341904	245 2 5 3 5 3 5 3 5 5 5 5 5 5 5 5 5 5 5 5	288 304 318 398 398 398 372	210 250 22h 2hu 2ho 260 2h7 27h 2h8 278 2h7 277 2h1 270 2h1 270 2h2 267	287 278 2766 5127 5512 5512 3664 2172 3664 149 501 509 506 217	161 165 166 165 155 155 160 161 165 165 165 165 165 165 165 165 165
								mixture,		t by	volume	ethyl	ene glyco		-E-2)	1 2					·
142 143 144 145 146 149 150 151 152 153 155 156	0.0.50.00 5.00 5.00 5.00 5.00 5.00 5.00	2200 2200 2200 2200 2200 1800 1800 1800	73 775 776 777 775 775 778 779 813	30.28 30.28 30.30 30.30 30.39 30.28 30.28 30.16 30.11 30.11 30.11	6722 6722 6722 6726 6735 500 500 550 550 550 555 555 555 555 5	83970 83370 83370 83340 63310 63270 10100 103370 10350	0.0801 .0801 .0803 .0805 .0806 .0802 .0802 .0809 .0807 .0805 .0805 .0805	>5 <b>to</b> 15<	208 2468 1851 2311 1868 247 2709 1911 2319 268	11.0 9.6 12.6 11.9 11.3 11.9 10.5 11.7 13.3 11.6 10.6	8.7 10.4 9.5 11.3 9.8 11.4 10.6 9.6	10.6 9.1 10.6 10.4 11.6 9.7 10.3 12.4 11.2 12.0 10.7	213 252 272 191 236 236 195 272 252 183 214 197 254 273	29.76.3.48 29.96.23.79.0 29.97.90 23.97.90 23.5.4.5.7.51 32.7.51	26.0 26.4 21.5 20.9 21.7 21.6 20.8 29.3	26.1 42.46.66.3.77.53.8 55.54.45.66.3.77.53.8 61.22.96.66.3.77.53.8				3841 4020 3318 3311 33560 4854 45450	

MR

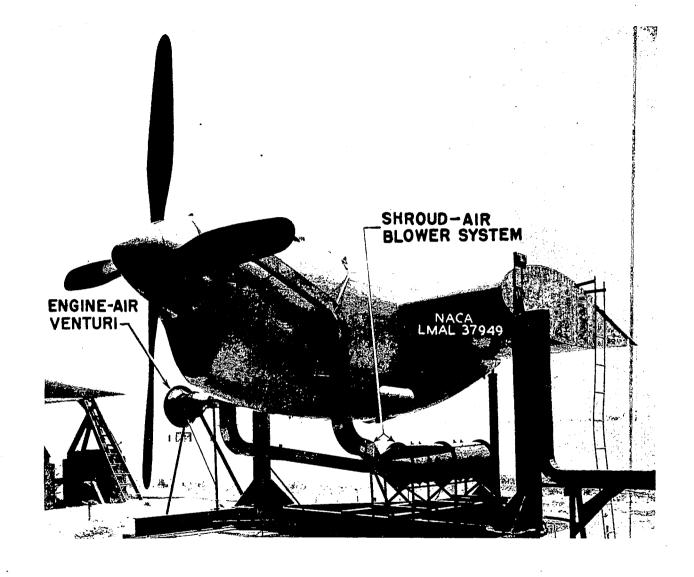


Figure 1.- Engine-nacelle test setup for cooling tests of the Allison V-3420-l1 engine.

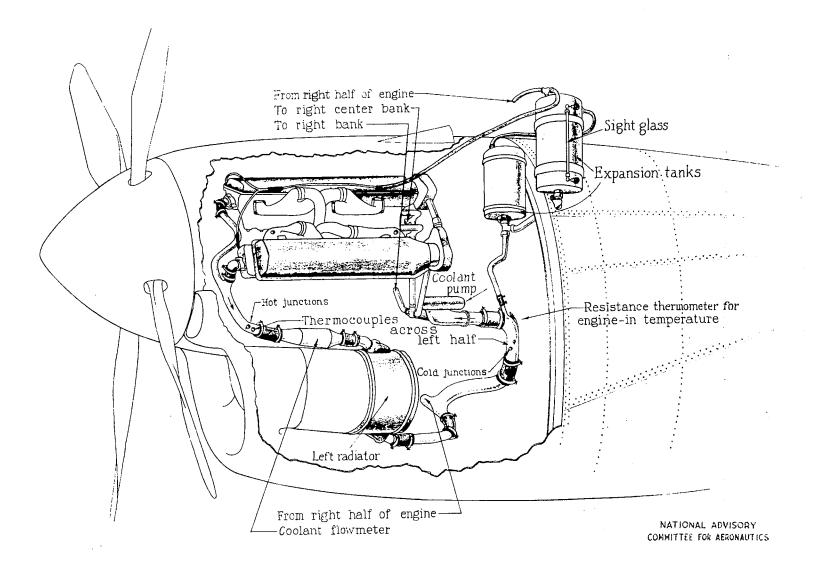


Figure 2.-Location of coolant flowmeter and thermocouples for left half of engine.

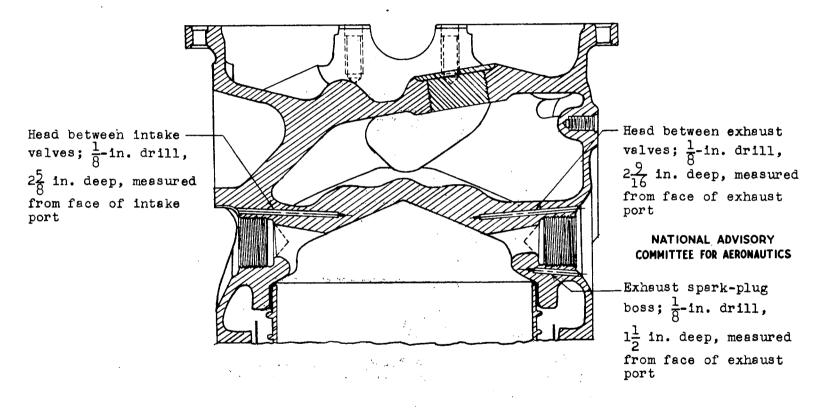
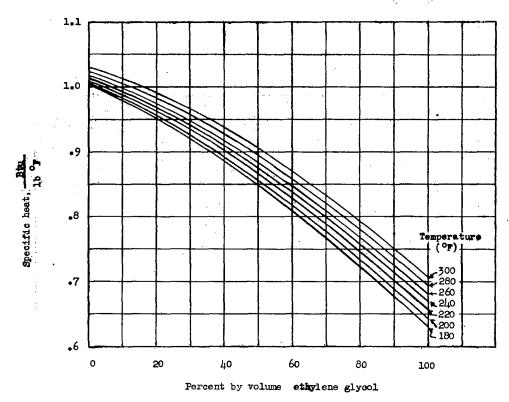
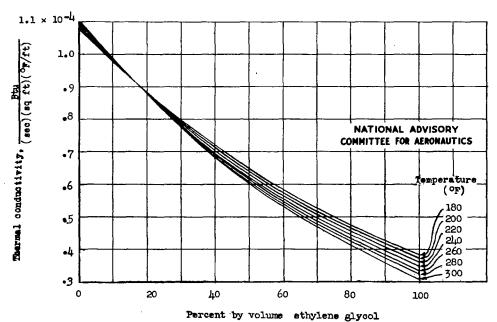


Figure 3.- Location of embedded thermocouples in cylinder wall.

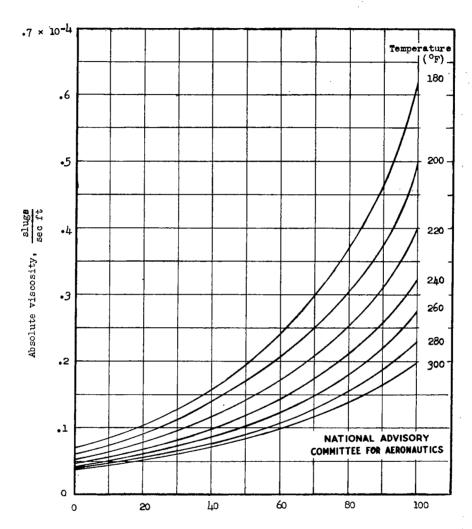


(a) Specific heat.



(b) Thermal conductivity.

Figure 4.- Physical properties of mixtures of pure ethylene glycol and water. (From reference 4.)



Percent by volume ethylene glycol

(c) Absolute viscosity.

Figure 4.- Concluded.

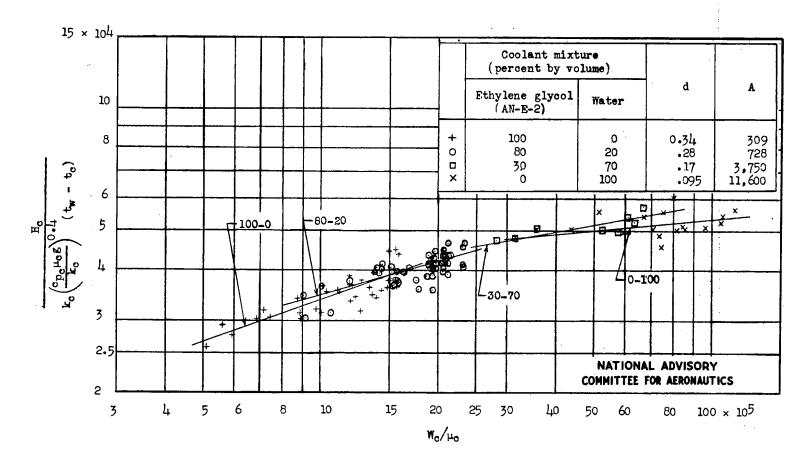


Figure 5.- The effect of  $W_c/\mu_c$  on the rate of heat transfer from the cylinder walls to the coolant.

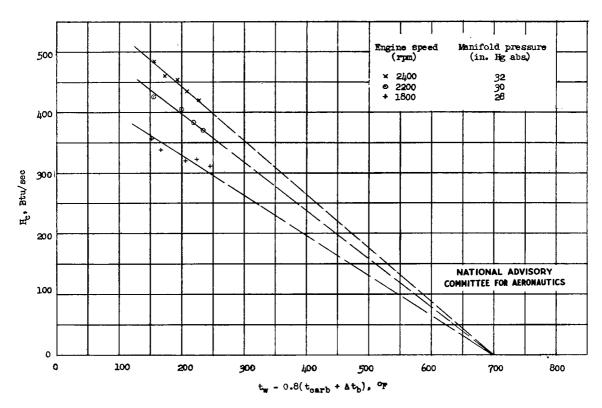


Figure 6.- The variation of the heat transfer from the cylinder walls to the coolant with the average cylinder-wall temperature; fuel-air ratio, 0.08.

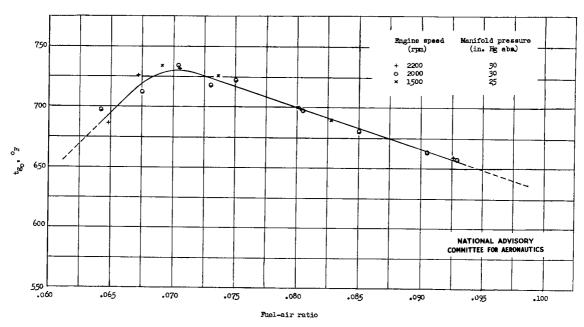


Figure 7.- The variation of  $t_{g_0}$  with fuel-air ratio.

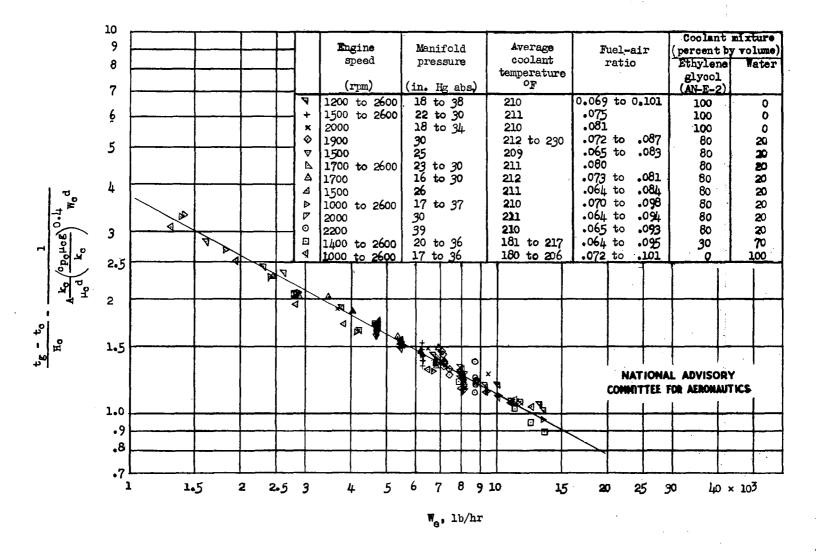
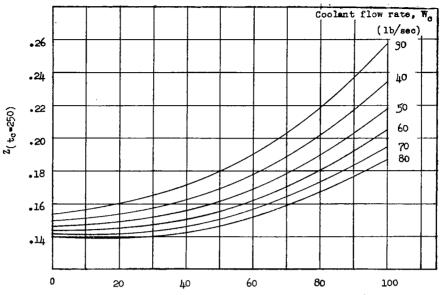
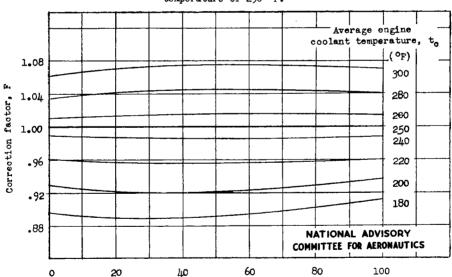


Figure 8.- Heat-rejection characteristics of the Allison V-3420-11 engine.



Percent by volume ethylene glycol (AN-E-2)

(a) Values of Z for an average engine coolant temperature of 250° F.

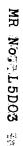


Percent by volume ethylene glycol (AN-E-2)

(b) Correction factor to be applied to Z<sub>(t<sub>c</sub>=250)</sub> to obtain Z for average engine coolant temperatures other than 250° F.

Figure 9.- Curves for determining Z or  $\frac{1}{\frac{k_c}{\mu_c d} \binom{c_{p_c} \mu_c g}{k_c}^{0 - l_t}} \text{ for various}$ 

coolant mixtures, average engine coolant temperatures, and coolant flow rates. Z = FZ(t<sub>o</sub>=250)\*



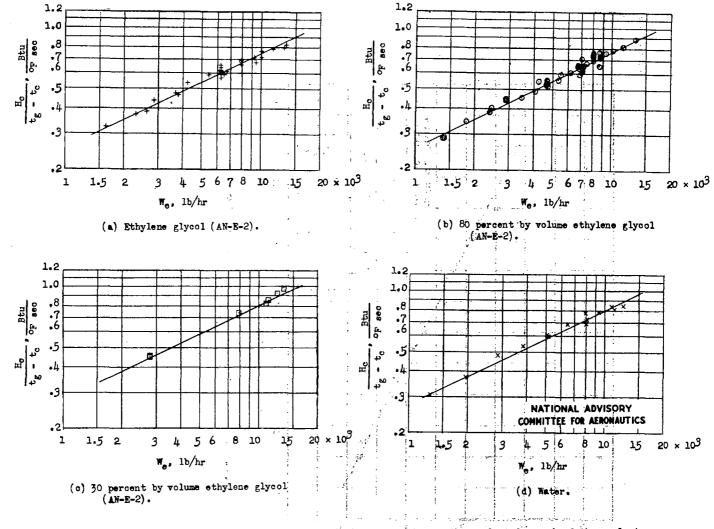


Figure 10.- The effect of engine-air flow on the heat rejection to the coolant for each of the coolant mixtures used in the tests. Effect of variation in coolant temperature on coolant properties neglected.

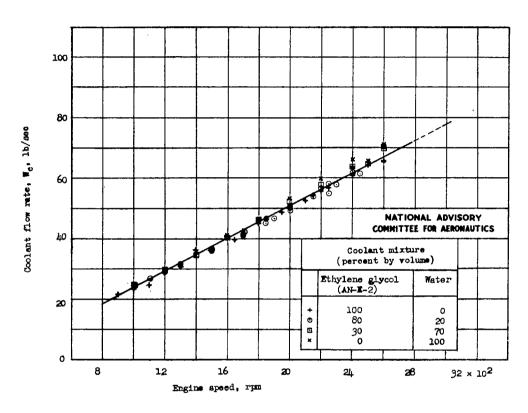


Figure 11.- The effect of engine speed on the coolant flow rate for various coolant mixtures.

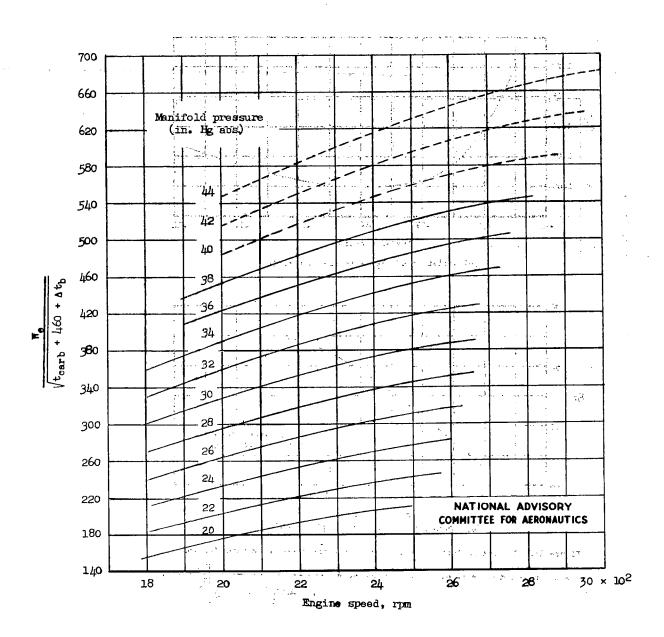
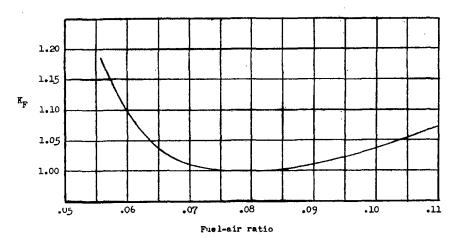
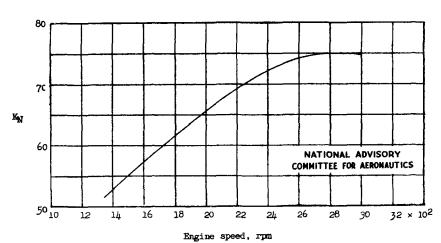


Figure 12.- Variation of engine-air flow with engine operating conditions. Exhaust back pressure, 30 inches of mercury absolute.



(a) Variation of Kp with fuel-air ratio.



(b) Variation of  $K_N$  with engine speed.

Figure 13.- Correction factors used to calculate the brake horsepower for the Allison V-3420-11 engine. Brake horsepower =  $\begin{bmatrix} & p_m & \\ & & \\ 1 + & \frac{t_{oarb} - 80}{10} & (0.01) \end{bmatrix}^{\frac{K_N}{K_F}} - 700.$ 

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